This article was downloaded by: On: 23 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713617200

Stereochemical Study of Cycloadditions Using Erythrose and Threose Based Dienes as Source of 2-Nonulosonic Acid Analogs

André Lubineau^a; Hélène Arcostanzo^a; Yves Queneau^a ^a Laboratoire de Chimie Organique Multifonctionnelle CNRS, URA n°462, Institut de Chimie Moléculaire d'Orsay Université Paris-Sud, Orsay Cédex, France

To cite this Article Lubineau, André , Arcostanzo, Hélène and Queneau, Yves(1995) 'Stereochemical Study of Cycloadditions Using Erythrose and Threose Based Dienes as Source of 2-Nonulosonic Acid Analogs', Journal of Carbohydrate Chemistry, 14: 9, 1307 — 1328

To link to this Article: DOI: 10.1080/07328309508005413 URL: http://dx.doi.org/10.1080/07328309508005413

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

STEREOCHEMICAL STUDY OF CYCLOADDITIONS USING ERYTHROSE AND THREOSE BASED DIENES AS SOURCE OF 2-NONULOSONIC ACID ANALOGS

André Lubineau, Hélène Arcostanzo and Yves Queneau*,1

Laboratoire de Chimie Organique Multifonctionnelle CNRS, URA n°462 Institut de Chimie Moléculaire d'Orsay Université Paris-Sud, Bât. 420 91405 Orsay Cédex, France

Received April 4, 1995 - Final Form July 17, 1995

ABSTRACT

The stereochemical outcome of the hetero Diels Alder reaction of an erythrose based diene with sodium glyoxylate was rationalized by preparing the same compounds via decarboxylation of 2-carbethoxy-2-deoxy-2-ulosonic esters, obtained by cycloaddition with diethyl ketomalonate. Further chemical transformations of cycloadducts allowed us to prepare a series of new 2-nonulosonic acid derivatives.

INTRODUCTION

The aqueous hetero Diels Alder reaction has been shown to be a useful methodology.² Reaction of glyoxylic acid as dienophile has been reported to be an efficient way to produce bicyclic α -hydroxy lactones,³ and reaction of conjugated dienes with a number of α -activated carbonyl compounds allowed for preparation of substituted dihydropyrans directly from their commercial aqueous solution.⁴ The reaction has been used to access the 2-deoxy-2-ulosonic acid skeleton in which the C-2-C-3 and C6-O-6 bonds were created in the cycloaddition step, providing new syntheses of KDO and some of its

analogs,⁵ as well as a short synthesis of racemic 2-heptulosonic acid derivatives.⁶ The present report describes a study of the reaction based on the same strategy in the nine-carbon series starting from an erythrose based diene.⁷ The structural study of 2-carbethoxy-2-deoxy-2-nonulosonates obtained by cycloaddition of various dienes with diethyl ketomalonate, made it possible to determine the streochemical outcome of the reaction. Further transformations of adducts obtained either from cycloaddition with glyoxylate (2-deoxy) and ketomalonate (2-carbethoxy-2-deoxy) provide new heavily substituted 2-nonulosonic acid analogs.⁸

RESULTS AND DISCUSSION

The cycloaddition in water (2.5 d at reflux) of (*E*)-hepta-4,6-diene-1,2,3-triol (1), having the D-erythro configuration, with sodium glyoxylate (2) provided a mixture of four adducts **3ab** and **4ab** that arise from endo and exo transition states and attack of the dienophile on both faces of the chiral diene (Scheme 1). These adducts were protected as their methyl tri-O-acetyl esters whose configurations were later determined as **5ab** and **6ab** (28:31:16:25, endo:exo = 44:56, Re:Si = 41:59),⁹ and isolated in a global 67% yield from 1. These results are compared to the low 25% yield obtained from the reaction of acetylated diene **7** with methyl glyoxylate (38:27:13:22, endo:exo = 65:35, Re:Si = 40:60). The high temperature required for the reaction with methyl glyoxylate (4 h, 140 °C) was not compatible with the stability of both the diene and the products, giving an additional example of the usefulness of using aqueous sodium glyoxylate as dienophile. It has to be noted that the reaction of diene **1** with sodium glyoxylate occurred faster in water than in other polar solvents such as ethanol or dimethylformamide.

In order to ascertain the structure of each of the four adducts produced in the reaction, the same compounds were prepared by cycloaddition of diene 7 with diethyl ketomalonate that provided adducts 10 and 11 (Re:Si = 35:65), which were decarboxylated and reprotected separately in 51% yield from 7. This allowed us to divide the four adducts in two groups, endo and exo Si (5ab) and endo and exo Re (6ab). The absolute configuration at C-6 was determined after careful structural study of adducts obtained from various dienes (*vide infra*).

In order to study the influence of the hydroxyl group protection on the facial selectivity, we prepared variously substituted dienes (Scheme 2). Thus,



a. 1. CH₃I, DMF 2. Ac₂O, Py 67% from 1



4. Ac₂O, Py 51% from **7**

SCHEME 1



a. BrPh₃PAll, BuLi, THF, 45%; b. Ph₃P=CHCHO, THF, 76%; c. Ph₃P=CH₂, THF, 76%; d. Ac₂O, py, 92%; e. NaH, BnBr, THF, 70%. f. 80% aq AcOH, 60 °C, 30 min, 90%; g. Ac₂O, py, quant.; h. reference 11.

reaction of 2,4-di-O-benzylidene-D-erythrose $(12)^{10}$ with allyltriphenylphosphorane in THF gave a 4:1 mixture of E and Z dienes (56%) out of which the major E diene 13^{11} crystallized. This reaction required two equivalents of phosphorane because of the presence of a free hydroxyl group in aldehyde 12. An alternative route to diene 13 consisted in two consecutive Wittig reactions, first with the stabilized triphenylphosphoranylideneacetaldehyde giving E enal 14 (76%) which provided diene 13 after methylenation (76%). Careful acidic hydrolysis of the acetal linkage of diene 13 (80% aq AcOH, 60 °C, 30 min) gave diene 1 in 90% yield;¹² increased reaction time led to degradation of the diene. Dienes 15 and 16 were obtained from 13 by standard procedures. Triacetylated diene 7 could either be obtained from acetylation of 1 or following a known procedure from tri-O-acetyl-D-glucal (17).^{11,13}

The same sequence starting from tri-O-acetyl-D-galactal provided diene **19**, epimer of **7** at C-7, having the D-*threo* configuration.

The outcome of the cycloaddition for the six dienes was studied using diethyl ketomalonate. Adducts **20** to **29** were obtained in fair to good yields (Scheme 3).¹⁴ As shown in Table 1, the substitution on hydroxyl groups of erythrose diene had little effect on the facial selectivity that varied between a 2:1 and a 1:1 ratio. Ratio for the mixture **20** + **21**, arising directly from unprotected diene **1**, was determined, after acetylation (Ac₂O, Py), upon comparison (¹H NMR study of crude mixture) with adducts **10** and **11**, arising from acetylated diene **7**. Mixtures **22** + **23** and **26** + **27**, obtained from benzylidene dienes **13** or **16**, were hydrogenated (5 atm. H₂, Pd/C) and acetylated (Ac₂O, Py) for comparison with fully reduced compounds **30** + **31** (cf. Scheme 4) arising from adducts **10** and **11** after double bond reduction (1 atm. H₂, Pd/C). Finally, mixture of adducts **24** + **25** arising from acetylated benzylidene diene **15** was compared with an acetylation mixture of **22** + **23**.

Comparison of the results obtained for dienes 7 and 19, epimeric at the allylic stereogenic center and leading to a reversal of the diastereofacial selectivity, allowed the determination of the absolute configuration at C-6 in adducts 10 and 11. Indeed, as depicted in Scheme 4, H-6-H-7 coupling constants in hydrogenated adducts 30 and 31 arising from "erythrose" adducts 10 and 11, and 32 and 33 arising from "threose" adducts 28 and 29, were consistent with the general trend in such structure, namely a small $J_{6,7}$ (0-2 Hz) in KDN or sialic acid type chairs, and a large $J_{6,7}$ (8-10 Hz) in KDO type chairs.¹⁵

Taking into account the low facial selectivity observed in these cycloadditions, this reaction has to be considered as an access to both series of dihydropyrans epimers at C-6, each of them being intermediates to interesting targets. Having in hand a range of 2,6-disubstituted dihydropyrans, and relying on our previous work on the functionalization of 2-deoxy derivatives in the heptulosonic⁶ and octulosonic⁵ acid series, we studied the further chemical elaboration of some of the newly made nonulosonate esters. Notably, adduct 11, having the same stereochemistry at position 6,7 and 8 as in KDN, was an interesting candidate for further functionalization, as depicted in Schemes 5 and 6. Thus, bishydroxylation of 11 via catalytic osmylation produced *cis* diol 34 in 81% yield, with total facial selectivity that assured correct stereochemistry at position 5. An oxidation-reduction sequence could permit inversion of the hydroxyl group at C-4. This step was achieved directly using diol 34 without further protection employing the stannylene methodology via the 4-hydroxy-3-ketone 35 which



Table 1. Facial selectivity for cycloadditions with diethyl ketomalonate.

Diene	conditions (°C, d)	yield (%)	products	Re:Si
1	60 (5)	57	20 + 21	30:70
1	80 (0.7)	85	20 + 21	40:60
7	10(1)	83	10 + 11	35:65
13	10(1)	72	22 + 23	40:60
15	10(1)	57	24 + 25	52:48
16	10(1)	87	26 + 27	55:45
19	10 (1)	50	28 + 29	63:37



SCHEME 4



a. cat OsO₄-NMO, Me₂CO-H₂O, r.t., 6 h, 81%; b. 1) Bu₂SnO, tol, Dean-Stark, 4 h; 2) Br₂, CH₂Cl₂, r.t., 80%; c. NaBH₄, CeCl₃, MeOH, r.t., 91% (**36:34** = 4:1); d. 1)LiOH, H₂O, r.t., 1 d; 2) pH2, 100 °C, 8 h; 3) pH 7-8 then CH₃I, DMF, r.t., 2 d; 4) Ac₂O, Py, r.t., 15 h, 73%.

was selectively obtained in 80% yield by bromine oxidation of stannylene derivative of **34**.¹⁶ Reduction of **35** under Luche conditions¹⁷ gave the desired trans diol **36** in 73% yield along with 18% of cis diol **34**. Compound **34** is a protected 2-deoxy-2-carbethoxy KDN derivative which could be decarboxylated and reprotected to provide a 2-deoxy KDN derivative (**37**) in 73% yield.

Activation of anomeric center was achieved using the methodology developed in the heptulosonic series,⁶ namely a *N*-bromosuccinimide mediated dibromination at C-2 and C-3 (Scheme 6). Methyl tetra-*O*-acetyl-2,3-dibromo-2,3-dideoxy- β -D-*erythro*-L-*manno*-2-nonulosonate (**38**) was thus obtained in 76% yield. ¹⁸ Dibromides **39** + **42** were obtained in 59% yield from the mixture **5ab** + **6ab**. The orientation of substituents was again shown to have a strong influence on the outcome of this reaction. For example, adducts **6ab** led to a low yield of dibromide **39** (C-4 epimer of **38**), while reaction of triacetates **41ab** (obtained from **5ab** together with **40**) gave dibromide **42** in a satisfactory 73% yield.



a. excess NBS, CCl₄, reflux, 5 d, 76%; b. 1) cat OsO_4 -NMO, Me₂CO-H₂O, r.t., r.t., 3 h, 82%; 2) same as a., 1 d, 34%; c. 1) same as b., 3 h, 90%; d. same as a., 7 h, 76%.

In conclusion, we have shown, that the hetero Diels Alder reaction, creating C-2-C-3 and the C-6-O-6 bonds, and subsequent dihydroxylation at C-4 and C-5, starting from an erythrose based diene, is a valuable route for preparation of new 2-deoxy-2-nonulosonic acid analogs. Stereochemistry of adducts at C-6 was established by preparing 2-carbethoxy-2-deoxy-2nonulosonates, that are formal C-glycosides of 2-nonulosonic acid, by reaction of various dienes with diethyl ketomalonate and subsequent decarboxylation. Further chemistry on either 2-carbethoxy-2-deoxy- or 2deoxy-2-nonulosonate provided a range of new 2-nonulosonic acid derivatives, including new protected KDN analogs.

EXPERIMENTAL

General. NMR spectra were recorded with Brüker AM 250 and 400 and AC 200 and 250 spectrometers. Chemical shifts are given in ppm downfield from internal tetramethylsilane; signal multiplicity is indicated as follows: s for singlet, d for doublet, t for triplet, q for quadruplet, m for multiplet and br for broad. IR spectra were recorded using a Brüker FT instrument. Flash-chromatography was performed using 6-35 μ silica gel (60) purchased from S.D.S. company. TLC was run using Merck 60 F254 plates, and visualized first with UV light and second by heating after alcoholic sulfuric or phosphomolybdic acid treatment. Melting points were measured on a Reichert apparatus and are uncorrected. Elementary analyses were performed at the "Service Central de Microanalyse du C.N.R.S."

(2-R,3-S)-1,3-O-Benzylidene-hepta-4,6-diene-1,2,3-triol (13). To a solution of allyltriphenylphosphonium bromide (1.42 g, 5 mmol) in THF (20 mL) was added dropwise *n*-butyllithium (1.6 M in hexane, 3.13 mL, 5 mmol). The orange-red solution was then warmed up to 55 °C, and a solution of 2,4benzylidene-D-erythrose¹⁸ **12** (416 mg, 2 mmol) in THF (20 mL) was added. After 5 min, the mixture was cooled to room temperature, diluted with ether and washed with brine. The organic layer was concentrated under vacuum and flash-chromatography of the residue (3:7 AcOEt-hexane) gave a mixture of diene 13 and its Z isomer in a 83:17 ratio (270 mg, 58%). Crystallization (AcOEt-hexane) gave the pure E diene 13 as white needles: mp 89-90 °C; $[\alpha]_{D^{20}}$ -103° (c 1.6, CH₂Cl₂) [lit.¹¹ mp 88-89 °C, $[\alpha]_{D^{20}}$ -106° (CHCl₃)]. ¹H NMR (CDCl₃, 200 MHz) δ 1.91 (brs, 1 H, OH), 3.56-3.72 (m, 2 H, H_{6.7ax}), 4.07 $(brt, J = 7 Hz, 1 H, H_5), 4.33 (dd, J = 16, 10 Hz, 1 H, H_{7eq}), 5.18 (dd, J = 10, 1)$ Hz, 1 H, H₁), 5.30 (dd, J = 16, 1 Hz, 1 H, H₁'), 5.54 (s, 1 H, PhCH), 5.80 (dd, J= 14, 7 Hz, 1 H, H₄), 6.27-6.53 (m, 2 H, H_{2.3}), 7.30-7.56 (m, 5 H, Ph); ^{13}C NMR (CDCl₃, 50 MHz) δ 70.69 (C₇), 65.31, 82.45 (C_{5.6}), 100.87 (PhCH), 119.00 (C₁), 126.10, 128.20, 128.96, 139.30, 134.65, 135.87, 137.36(C_{2,3,4},Ph).

Anal. Calcd for $C_{14}H_{16}O_3$: C, 72.39; H, 6.94; O, 20.66. Found: C, 72.53; H, 6.81; O, 20.85.

Preparation of diene 13 via aldehyde 14. 1) Preparation of aldehyde 14: to a solution of aldehyde 12^{19} (416 mg, 2 mmol) in THF (10 mL) was added portionwise (triphenylphosphoranylidene) acetaldehyde (670 mg, 2.2 mmol). After 6 h at room temperature, the solvent was evaporated, and flashchromatography (9:1 toluene-acetone) of the residue gave aldehyde 14 (357 mg, 76%). 2) Preparation of diene 12 : to a solution of aldehyde 14 (117 mg, 0.5 mmol) in THF (2 mL), a salt-free 1 M solution of methylenetriphenylphosphorane in toluene (1.1 mL) was added dropwise. The mixture was immediately poured in a CH₂Cl₂ / pH7 phosphate buffer mixture. The organic layer was concentrated and flash-chromatography of the residue (1:3 AcOEt-hexane) gave diene **13** (88 mg, 76%) identical with an authentic sample.

(2-*R*,3-*S*)-2-*O*-Acetyl-1,3-*O*-benzylidene-hepta-4,6-diene-1,2,3-triol (15). A solution of diene 13 (464 mg, 2 mmol) in acetic anhydride (2 mL) and pyridine (2 mL) was stirred overnight at room temperature. Evaporation of solvents and coevaporation with toluene followed by flash-chromatography of the residue (AcOEt-hexane, 1:4) gave diene 15 (507 mg, 92%) which crystallized from hexane: mp 85-86 °C, $[\alpha]_D^{20}$ -63° (*c* 1.9, CH₂Cl₂) [lit.11 mp 78-80 °C, $[\alpha]_D^{20}$ -58° (CHCl₃)]. 1H NMR (CDCl₃, 200 MHz) δ 2.00 (s, 3H, CH₃) 3.63 (t, *J* = 10 Hz, 1 H, H_{7ax}), 4.23 (dd, *J* = 9.5, 6 Hz, 1 H, H₅), 4.37 (dd, *J* = 10, 5 Hz, 1 H, H_{7eq}), 4.84 (dt, *J* = 9.5, 9.5, 5 Hz, 1 H, H₆), 5.12 (dd, *J* = 10, 1 Hz, 1 H, H₁), 5.23 (dd, *J* = 16, 1 Hz, 1 H, H₁'), 5.53 (s, 1 H, PhCH), 5.71 (dd, *J* = 14, 6 Hz, 1 H, H₄), 6.20-6.44 (m, 2 H, H_{2,3}), 7.30-7.55 (m, 5 H, Ph); ¹³C NMR (CDCl₃, 50 MHz) δ 20.62 (CH₃), 67.96 (C₇), 66.21, 79.57 (C_{5,6}), 101.05 (PhCH), 118.71 (C₁), 126.08, 128.13, 128.59, 128.94, 134.29, 135.88, 137.14 (C_{2.3.4}, Ph), 169.46 (C=O).

Anal. Calcd for C₁₆H₁₈O₄: C, 70.05; H, 6.61; O, 23.33. Found: C, 70.08; H, 6.82; O, 23.30.

(2-R,3-S)-2-O-Benzyl-1,3-O-benzylidene-hepta-4,6-diene-1,2,3-triol (16). To a solution of diene 13 (232 mg, 1 mmol), benzyl bromide (0.14 mL, 1.2 mmol) in THF (2 mL) was added sodium hydride (60% in mineral oil, 48 mg, 1.2 mmol). After the reaction mixture was strirred for 4 h at 50 $^{\circ}$ C, excess sodium hydride was destroyed by addition of water and the product was extracted with CH_2Cl_2 . The organic layer was concentrated and flashchromatography of the residue (1:9 AcOEt-hexane) gave diene 16 (225 mg, 70%) which crystallized neat when refrigerated: mp 69-71 °C; $[\alpha]_D^{20}$ -80° (c 1.4, CH_2Cl_2). ¹H NMR (CDCl₃, 200 MHz) δ 3.44 (ddd, J = 9.5, 8, 4 Hz, 1 H, H₆), 3.64 (t, J = 10 Hz, 1 H, H_{7ax}), 4.18 (dd, J = 8, 6 Hz, 1 H, H₅), 4.30 (dd, J= 10, 4 Hz, 1 H, H_{7eq}), 4.47-4.61 (AB system, 2 H, CH₂Ph), 5.13 (dd, J = 9, 1 Hz, 1 H, H₁), 5.24 (dd, J = 16, 1 Hz, 1 H, H₁'), 5.50 (s, 1 H, PhCH), 5.83 (dd, J= 14, 6 Hz, 1 H, H₄), 6.25-6.52 (m, 2 H, H_{2.3}), 7.30-7.55 (m, 5 H, Ph); ^{13}C NMR (CDCl₃, 50 MHz) δ 69.50 (C₇), 72.79, 80.70 (C_{5.6}, CH₂Ph), 100.81 (PhCH), 118.22 (C₁), 126.10, 127.93, 128.17, 128.41, 128.87, 130.04, 133.53, 136.27, 137.58, 137.68 (C_{2,3,4}, 2Ph).

Anal. Calcd for C₂₁H₂₂O₃: C, 78.23; H, 6.88; O, 14.89. Found: C, 77.99; H, 6.82; O, 14.93.

(2-*R*,3-*S*)-Hepta-4,6-diene-1,2,3-triol (1). A solution of diene 13 (1.74 g, 7.5 mmol) in acetic acid and water (8:2, 20 mL) was stirred 35 min at 60 °C. Concentration to dryness followed by flash-chromatography of the residue (CH₂Cl₂-MeOH, 9:1) gave diene 1 (971 mg, 90%); mp 71-72 °C, $[\alpha]_D^{20}$ -41° (*c* 1.4, MeOH). ¹H NMR (CD₃OD, 200 MHz) δ 3.46-3.73 (m, 3 H, H_{6,7,7}), 4.10 (dd, *J* = 6.5, 4 Hz, 1 H, H₅), 5.08 (dd, *J* = 9.5, 1 Hz, 1 H, H₁), 5.20 (dd, *J* = 16, 1 Hz, 1 H, H₁), 5.82 (dd, *J* = 14, 6.5 Hz, 1 H, H₄), 6.20-6.50 (m, 2 H, H_{2,3}); ¹³C NMR (CD₃OD, 50 MHz) ∂ 64.25 (C₇), 73.91, 75.98 (C_{5,6}), 117.48 (C₁), 133.48, 134.30, 137.86 (C_{2,3,4}).

Anal. Calcd for C₇H₁₂O₃: C, 58.32; H, 8.39; O, 32.29. Found: C, 58.19; H, 8.28; O, 32.21.

(2-*R*,3-*S*)-1,2,3-Tri-*O*-acetyl-hepta-4,6-diene-1,2,3-triol (7). Acetylated diene 7 could be obtained either by quantitative acetylation of diene 1 (Ac₂O, pyridine, r.t.) or following the procedure described in ref. 11 from tri-*O*-acetyl-D-glucal; $[\alpha]_D^{20} + 34^{\circ}$ (*c* 2.4, CH₂Cl₂),+28° (*c* 3, CHCl₃) [lit.¹¹ +25° CHCl₃). ¹H NMR (CDCl₃, 250 MHz) δ 2.04 (s, 3 H, CH₃), 2.07 (s, 6 H, 2 CH₃), 4.16 (dd, *J* = 12, 6.5 Hz, 1 H, H₇), 4.25 (dd, *J* = 12, 4 Hz, 1 H, H₇), 5.13-5.36 (m, 3 H, H_{1,1',6}), 5.51 (dd, *J* = 7, 4.5 Hz, 1 H, H₅), 5.65 (dd, *J* = 14, 7 Hz, 1 H, H₄), 6.21-6.42 (m, 2 H, H_{2,3}); ¹³C NMR (CDCl₃, 50 MHz) δ 20.57, 20.73, 20.82 (3 CH₃) 61.75 (C₇) 71.45, 72.12 (C_{5,6}) 119.65 (C₁), 125.96, 135.33, 135.64 (C_{2,3,4}), 169.48, 169.93, 170.42 (3 C=O).

Anal. Calcd for C₁₃H₁₈O₆: C, 57.77; H, 6.71; O, 35.52. Found: C, 57.74; H, 6.60; O, 35.26.

(2-R,3-R)-1,2,3-Tri-O-acetyl-hepta-4,6-diene-1,2,3-triol (19). The same method as for diene 7 following ref. 10 from 18^{18} gave diene 19 in a 44% yield; $[\alpha]_D^{20} + 16^{\circ} (c \ 1.2, CH_2Cl_2)$. 1H NMR (CDCl₃, 200 MHz) δ 2.07 (s, 3 H, CH₃), 2.10 (s, 6 H, 2 CH₃), 4.03 (dd, J = 12, 6 Hz, 1 H, H₇), 4.34 (dd, J = 12, 4 Hz, 1 H, H₇), 5.18-5.35 (m, 3 H, H_{1,1',6}), 5.48-5.64 (m, 2 H, H_{4,5}), 6.20-6.40 (m, 2 H, H_{2,3}); 13C NMR (CDCl₃, 50 MHz) δ 20.21, 20.30, 20.45 (3CH₃), 61.77 (C₇), 70.91, 71.49 (C_{5,6}), 119.30 (C₁), 125.90, 135.16 (C_{2,3,4}), 169.16, 169.51, 169.95 (3C=O).

Anal. Calcd for C₁₃H₁₈O₆: C, 57.77; H, 6.71; O, 35.52. Found: C, 58.06; H, 6.49; O, 35.40.

Cycloaddition of diene 1 with sodium glyoxylate. Preparation of adducts **5ab** and **6ab**. A solution of diene 1 (501 mg, 3.5 mmol) and sodium glyoxylate monohydrate (1.6 g, 14 mmol) in water (7 mL) was heated in a

closed tube at 120 °C in the presence of a small amount of hydroquinone. After 2.5 d, TLC (7:3 1-propanol-water) showed total disappearance of starting diene. The mixture was cooled to room temperature, diluted with water (10 mL total) and a 2:1 (vol.) mixture of DMF and CH_{3I} (15 mL) was added. After 4 d at room temperature, solvents were removed under vacuum and the mixture was dissolved in a 1:1 (vol) mixture of Ac₂O and pyridine (20 mL) and left overnight at room temperature. Removal of excess reagents by coevaporation several times with toluene gave a residue which was diluted with CH_2Cl_2 (50 mL). After washings (water, 50 mL, brine, 50 mL), the solvent was evaporated and the residue was purified by flashchromatography (1:3 AcOEt-hexane) to give mixture of adducts **5ab** and **6ab** (850 mg, 68%) in a 44:56 endo:exo ratio and a 41:59 Re:Si ratio as determined by NMR analysis and comparison with same compounds prepared from **10** and **11**.

General procedure for the cycloadditions using diethyl ketomalonate. Cycloaddition of diene 7 with diethyl mesoxalate. Preparation of adducts 10 and 11. Diene 7 (1.35 g, 5 mmol) and commercial diethyl mesoxalate (3.05 mL, 20 mmol) were heated at 100 °C in the presence of a catalytic amount of hydroquinone during 24 h. The mixture was then cooled, diluted with CH_2Cl_2 and washed with water (2x50 mL). After evaporation of the solvent, preparative HPLC of the residue (AcOEt-hexane, 1:3 then 1:2) gave in order of elution first the unreacted diene followed by the partially separated adducts 10 and 11 (1.592 g) in a 83% yield (based on starting diene recovery, 72% isolated yield). Pure fractions of each adduct were characterized.

Ethyl 2-Carbethoxy-2,3,4,5-tetradeoxy-D*ribo*-non-4-en-2ulosonate (10). $[\alpha]_D^{20}$ +25° (*c* 2.5, CH₂Cl₂). ¹H NMR (CDCl₃, 200 MHz) δ 1.27, 1.28 (2 t, 3 H each, 2 CH₃), 2.04 (s, 3 H, CH₃), 2.06 (s, 3 H, CH₃), 2.12 (s, 3 H, CH₃) 2.61 (ddt, *J* = 17, 4, 3, 3 Hz, 1 H, H3), 2.82 (dddd, *J* = 17, 6, 3, 1 Hz, 1 H, H₃), 4.10-4.40 (m, 5 H, 2 CH₂, H₉), 4.62 (dd, *J* = 12, 2 Hz, 1 H, H9), 4.64-4.74 (m, 1 H, H₆), 5.19 (dd, *J* = 5, 4 Hz, 1 H, H7), 5.39 (ddd, *J* = 6, 4, 2 Hz, 1 H, H8), 5.69 (brd, *J* = 10 Hz, 1 H, H5), 5.92-6.04 (m, 1 H, H4); ¹³C NMR (CDCl₃, 50 MHz) δ 13.48, 13.60 (2 CH₃ Et), 20.31, 20.42 (3 CH₃ Ac), 28.50 (C₃), 61.64, 61.84, 62.43 (2 CH₂ Et, C9), 69.67, 71.08, 72.77 (C_{6,7,8}), 79.38 (C₂), 123.91, 124.06 (C_{4,5}), 167.04, 167.58, 169.42, 169.61, 170.16 (5 C=O).

Anal. Calcd for C₂₀H₂₈O₁₁: C, 54.05; H, 6.35; O, 39.60. Found: C, 53.78; H, 6.31; O, 39.67.

Ethyl 2-Carbethoxy-2,3,4,5-tetradeoxy-D-arabino-non-4-en-2ulosonate (11). $[\alpha]_D{}^{20}$ +58° (c 1.9, CH₂Cl₂). 1H NMR (CDCl₃, 200 MHz) δ 1.25, 1.28 (2 t, J = 7 Hz, 3 H each, 2 CH₃), 2.05 (s, 6 H, 2 CH₃), 2.10 (s, 3 H, CH₃), 2.58 (ddt, J = 17, 4, 3, 3 Hz, 1 H, H₃), 2.73 (dddd, J = 17, 6, 3, 1 Hz, 1 H, H₃'), 4.10-4.40 (m, 5 H, 2 CH₂, H₉), 4.46 (dd, J = 12.5, 2.5 Hz, 1 H, H₉'), 4.83-4.91 (m, 1 H, H₆), 5.32 (dd, J = 6.5, 2 Hz, 1 H, H₇), 5.41 (ddd, J = 6.5, 6, 2 Hz, 1 H, H₈), 5.57 (brd, J = 10 Hz, H₅), 5.87-5.99 (m, 1 H, H₄); ¹³C NMR (CDCl₃, 50 MHz) δ 13.63, 13.79 (2 CH₃ Et), 20.45, 20.59 (3 CH₃ Ac), 28.56 (C₃), 61.65, 61.91, 62.12 (2 CH₂ Et, C₉), 69.40, 70.72, 71.51 (C_{6,7,8}), 79.42 (C₂), 124.40, 124.65 (C_{4,5}), 167.04, 168.05, 169.53, 169.73, 170.38 (5 C=O).

Anal. Calcd for $C_{20}H_{28}O_{11}$, C : 54.05, H : 6.35, O : 39.60. Found, C : 53.77, H : 6.35, O : 39.62.

Cycloaddition of diene 1 with diethyl mesoxalate. Diene 1 (144 mg, 1 mmol) was heated with diethyl mesoxalate (0.31 mL, 2 mmol) in presence of a catalytic amount of hydroquinone at 80 °C for 17 h. Flash-chromatography of the mixture (95:5 CH₂Cl₂-MeOH) gave a 40:60 ratio of **20** and **21** (194 mg, 61%, 85% based on starting diene recovery), which were identified with **10** and **11** after quantitative peracetylation (Ac₂O-pyridine, 1:1).

2,3,4,5-Tetradeoxy-D-arabino-non-4-en-2-ulosonate Methyl (6ab). This reaction could be achieved either from 11 or 21. Adduct 21 (161 mg, 0.5 mmol) was treated overnight with aqueous LiOH (1M, 2 mL). When TLC (1-propanol-water, 7:3) indicated the disappearance of both ester fonctions, the mixture was acidified with Dowex-50X8 H+ (1.3 g), until the pH decreased to 2, and then heated at 100 °C. After 24 h, TLC showed that decarboxylation had occurred. The mixture was then cooled and filtered over celite. After evaporation of the solvent, the residue was dissolved in methanol, and treated with a freshly prepared solution of diazomethane in ether. After concentration, flash-chromatography of the residue (95:5 CH₂Cl₂-MeOH) gave a mixture (70 mg, 60%), which was quantitatively acetylated with a mixture of acetic anhydride and pyridine (1:1 vol, 2 mL). Partial separation of the two epimers by flash-chromatography (AcOEthexane, 1:3) allowed assignment of the following values in their NMR spectra. 6a: ¹H NMR (CDCl₃, 200 MHz) δ 2.06, 2.08, 2.10 (3s, 3 H each, 3 CH_3), 2.25-2.45 (m, 2 H, $H_{3,3'}$), 3.77 (s, 3 H, CO_2Me), 4.18 (dd, J = 10, 7 Hz, 1 H, H₂), 4.24 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₆), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.47-4.54 (m, 1 H, H₉), 4.59 (dd, J = 12.5, 6 Hz, 1 H, H₉), 4.59 (dd, J = 12.5, 6 12.5, 2.5 Hz, 1 H, Hg, $5.32 (dd, J = 6, 2.5 \text{ Hz}, 1 \text{ H}, \text{H}_7), 5.42 (dt, J = 6, 6, 2.5 \text{ Hz})$ Hz, 1 H, H₈), 5.53-5.67 (m, 1 H, H₅), 5.90-6.04 (m, 1 H, H₄). **6b**: ¹H NMR (CDCl₃, 250 MHz) δ 2.07, 2.09, 2.11 (3s, 3 H each, 3 CH₃), 2.25-2.53 (m, 2 H, H_{3,3}), 3.75 (s, 3 H, CO₂Me), 4.24 (dd, *J* =12.5, 5.5 Hz, 1 H, H₉), 4.41 (dd, *J* = 12.5, 3 Hz, 1 H, H₉), 4.54 (t, *J* = 5.5 Hz, 1 H, H₂), 4.66-4.74 (m, 1 H, H₆), 5.31 (dd, *J* = 7, 3.5 Hz, 1 H, H₇), 5.34-5.46 (m, 1 H, H₈), 5.57-5.73 (m, 1 H, H₅), 5.97 (ddt, *J* = 10, 4, 4, 2 Hz, 1 H, H₄); ¹³C NMR (CDCl₃, 50 MHz) δ 20.62 (3CH₃ Ac), 26.40 (C₃), 51.97 (CO₂Me), 61.89 (C₉), 69.23, 69.45, 70.52, 71.46 (C_{2,6,7,8}), 124.81, 125.61 (C_{4,5}), 169.70, 170.56, 171.62 (4 C=O).

Anal. Calcd for C₁₆H₂₂O₉ (mixture **6ab**): C, 53.63; H, 6.19. Found: C, 54.24; H, 6.24.

Methyl 2,3,4,5-Tetradeoxy-D-*ribo*-non-4-en-2-ulosonate (5ab). Employing the same decarboxylation procedure as above from isomer 10 lead to the mixture of peracetylated compounds **5a** and **5b**. The following NMR assignments have been made. **5a** : ¹H NMR (CDCl₃, 200 MHz) δ 2.04, 2.06, 2.12 (3s, 3 H each, 3 CH₃) 2.27-2.58 (m, 2 H, H_{3,3'}), 3.77 (s, 3 H, CO₂Me), 4.21 (dd, J = 10, 5 Hz, 1 H, H₂), 4.35 (dd, J = 12.5, 8 Hz, 1 H, H₉), 4.41-4.52 (m, 1 H, H₆), 4.58 (dd, J = 12.5, 3 Hz, 1 H, Hg), 5.13 (dd, J = 5, 4 Hz, 1 H, H₇), 5.35-5.46 (m, 1 H, H₈), 5.65-5.77 (m, 1 H, H₅), 5.92-6.05 (m, 1 H, H₄). **5b**: ¹H NMR (CDCl₃, 250 MHz) δ 2.05 (s, 6 H, 2 CH₃), 2.11 (s, 3 H, CH₃), 2.27-2.61 (m, 2 H, H_{3,3'}), 3.75 (s, 3 H, CO₂Me), 4.28 (dd, J = 12, 7.5 Hz, 1 H, H₉), 4.46 (dd, J = 12, 3 Hz, 1 H, H_{9'}), 4.56 (t, J = 5.5 Hz, 1 H, H₂), 4.63-4.73 (m, 1 H, H₆), 5.22 (dd, J = 6.5, 4 Hz, 1 H, H₇), 5.43 (ddd, J = 7.5, 4, 3 Hz, 1 H, H₈), 5.63-5.76 (m, 1 H, H₅), 5.87-6.02 (m, 1 H, H₄); ¹³C NMR (CDCl₃, 50 MHz) δ 20.69 (3CH₃ Ac), 26.09 (C₃), 51.99 (CO₂Me), 62.13 (C₉), 69.56, 69.81, 72.49 (C_{2,6,7,8}), 124.44, 124.73 (C_{4,5}), 169.60, 169.95, 170.50, 171.59 (4 C=O).

Anal. Calcd for C₁₆H₂₂O₉ (mixture **5ab**): C, 53.63; H, 6.19. Found: C, 53.89; H, 6.32.

Preparation of hydrogenated adducts **30** and **31.** Adducts **10** or **11** (1.5 mmol) were dissolved in ethyl acetate (20 mL) and 10% Pd/C (200 mg) was added. After 5 h of hydrogenation (1 atm) conditions, the mixture was filtered over celite to give adducts **30** and **31** (94 and 97% resp.) after concentration to dryness.

Ethyl 2-Carbethoxy-2,3,4,5-tetradeoxy-D-*ribo*-non-2-ulosonate (30). $[\alpha]_D^{20}$ +41° (c 2.4, CH₂Cl₂). ¹H NMR (CDCl₃, 200 MHz) δ 1.20-1.90, 2.31-2.43 (m, 12 H, 2 CH₃, H_{3,3',4,4',5,5'}), 2.05 (s, 3 H, CH₃), 2.07 (2 s, 6 H, 2 CH₃), 3.87 (ddd, J = 10.5, 7, 2 Hz, 1 H, H₆), 4.22, 4.28 (2 q, J = 7 Hz, 4 H, 2

CH₂), 4.45 (dd, J = 12.5, 8 Hz, 1 H, H₉), 4.55 (dd, J = 12.5, 3 Hz, 1 H, H₉), 5.08 (dd, J = 7, 3 Hz, 1 H, H₇), 5.49 (dt, J = 8, 3, 3 Hz, 1 H, H₈); ¹³C NMR (CDCl₃, 50 MHz) δ 13.33, 13.49 (2 CH₃ Et), 20.12, 20.26 (3 CH₃ Ac), 18.84, 26.02, 28.77 (C_{3,4,5}), 61.28, 61.43, 62.01 (2CH₂ Et, C₉), 69.95, 71.88, 73.14 (C_{6,7,8}), 81.24 (C₂), 167.19, 167.71, 169.12, 169.45, 169.94 (5 C=O).

Anal. Calcd for C₂₀H₃₀O₁₁: C, 53.80; H, 6.77; O, 39.42. Found: C, 54.08; H, 6.84; O, 39.39.

Ethyl 2-Carbethoxy-2,3,4,5-tetradeoxy-D-arabino-non-2ulosonate (31). $[\alpha]_D^{20}$ -1° (c 3, CH₂Cl₂). ¹H NMR (CDCl₃, 200 MHz) δ 1.20-1.90, 2.25-2.38 (m, 12 H, 2 CH₃, H_{3,3',4,4',5,5'), 2.04, 2.10, 2.14 (3 s, 3 H each, 3 CH₃), 4.05 (dt, J = 10.5, 2.5, 2.5 Hz, 1 H, H₆), 4.14-4.40 (m, 5 H, 2 CH₂, H₉), 4.54 (dd, J = 12.5, 2.5 Hz, 1 H, H₉'), 5.21 (dd, J = 6, 2.5 Hz, 1 H, H₇), 5.37 (dt, J = 6, 6, 2.5 Hz, 1 H, H₈); ¹³C NMR (CDCl₃, 50 MHz) δ 13.35, 13.53 (2CH₃ Et), 20.16,20.39 (3 CH₃ Ac), 19.26, 25.27, 28.59 (C_{3,4,5}), 61.19, 61.41, 61.80 (2CH₂ Et, C₉), 69.93, 71.73, 72.25 (C_{6,7,8}), 81.32 (C₂), 167.17, 167.84, 169.64, 169.74, 170.09 (5 C=O).}

Anal. Calcd for C₂₀H₃₀O₁₁: C, 53.80; H, 6.77; O, 39.42. Found: C, 53.67; H, 6.55; O, 39.17.

Preparation of compounds 28, 29, 32 and 33. Adducts 28 and 29 were obtained from diene 19 following the general procedure for cycloadditions using diethyl ketomalonate and subsequent hydrogenation (cf. preparation of compounds 30 and 31).

Ethyl 2-Carbethoxy-2,3,4,5-tetradeoxy-D-xylo and lyxo-non-4-en-2-ulosonate (28 + 29). ¹H NMR (CDCl₃, 200 MHz) δ 1.20-1.45 (m, 6 H, 2 CH₃), 1.98, 2.14, 2.18 (3 s, 3 CH₃₍₂₉₎), 1.99, 2.02, 2.05 (3 s, 3 CH₃₍₂₈₎), 2.54-2.85 (m, 2 H, H_{3,3}'), 4.05 (dd, J = 12, 6 Hz, H₉₍₂₉₎), 4.12-4.45 (m, 5 H, 2 CH₂₊ H_{9'}), 4.56 (dd, J = 12, 6 Hz, H₉₍₂₈₎), 4.62-4.78 (m, 1 H, H₆), 5.18 (dd, J = 7, 4 Hz, 1 H, H₇₍₂₉₎), 5.32 (dd, J = 7, 3 Hz, 1 H, H₇₍₂₈₎), 5.40-5.56 (m, 1 H, H₈), 5.57-5.68, 5.88-6.00 (2 m, 2 H, H_{4,5})

Anal. Calcd for C₂₀H₂₈O₁₁: C, 54.05; H, 6.31; O, 39.60. Found: C, 54.34; H, 6.42; O: 39.33.

Ethyl 2-Carbethoxy-2,3,4,5-tetradeoxy-D-xylo-non-2-ulosonate (32). ¹H NMR (CDCl₃, 200 MHz) δ 1.20-1.35 (m, 6 H, 2 CH₃), 1.40-1.90, 2.26-2.38 (m, 6 H, H_{3,3',4,4',5,5'}), 2.04, 2.07, 2.12 (3 s, 3 H each, 3 CH₃), 4.02 (dt, J = 11, 3, 3 Hz, 1 H, H₆), 4.16-4.35 (m, 4 H, 2 CH₂), 4.46 (dd, J = 12, 4 Hz, 1 H, H₉), 4.62 (dd, J = 12, 3 Hz, 1 H, H₉), 5.20 (dd, J = 7.5, 3 Hz, 1 H, H₇), 5.40

(ddd, J = 7.5, 4, 3 Hz, 1 H, H₈); ¹³C NMR (CDCl₃, 50 MHz) δ 13.80, 13.96 (2 CH₃ Et), 20.76 (3CH₃ Ac), 19.39, 25.59, 29.03 (C_{3,4,5}), 61.77, 61.93, 62.22 (2CH₂ Et, C₉), 70.51, 72.27, (C_{6,7,8}), 81.74 (C₂), 167.52, 168.59, 169.98, 170.35, 170.57 (5 C=O).

Ethyl 2-Carbethoxy-2,3,4,5-tetradeoxy-D-*lyxo*-non-2-ulosonate (33). ¹H NMR (CDCl₃, 200 MHz) δ 1.23-1.37 (m, 6 H, 2 CH₃), 1.37-1.90, 2.25-2.40 (m, 6 H, H_{3,3',4,4',5,5'}), 2.04, 2.10, 2.14 (3 s, 3 H each, 3 CH₃), 3.90 (ddd, J = 10, 8, 2 Hz, 1 H, H₆), 4.06-4.40 (m, 6 H, 2 CH₂ + H_{9,9'}), 5.14 (dd, J = 8, 4 Hz, 1 H, H₇), 5.48 (dt, J = 6, 6, 4 Hz, 1 H, H₈); ¹³C NMR (CDCl₃, 50 MHz) δ 13.75, 13.92 (2 CH₃ Et), 20.56, 20.72 (3 Ac), 19.28, 26.33, 29.30 (C_{3,4,5}), 61.55, 61.88, 63.17 (2CH₂ Et, C₉), 68.84, 71.58, 72.42 (C_{6,7,8}), 81.58 (C₂), 167.66, 168.02, 168.22, 169.87, 170.49 (5 C=O).

Ethyl 7,8,9-Tri-O-acetyl-2-carbethoxy-2,3-dideoxy-D-glycero-Dtalo-non-2-ulosonate (34). Compound 11 (1.6 g, 3.6 mmol) and N-methyl morpholine oxide (0.95 g, 7 mmol) were dissolved in a water-acetone mixture (1:8, 10 mL) and a 0.1 M solution of OsO4 in t-BuOH (2 mL, 0.2 mmol, 0.055 eq) was added dropwise. After 6 h of stirring at room temperature, saturated aqueous sodium hydrogenosulfite was added, and the mixture was extracted with AcOEt (2 x 80 mL). The organic layer was dried over sodium sulfate, filtered and concentrated to give quantitatively the crude diol 34 which was further purified after chomatography (AcOEt : hexane, 2:1, 1.4 g, 81%); $[\alpha]_D^{20} + 12^{\circ} (c \ 1.2, CH_2Cl_2)$. ¹H NMR (CDCl₃, 250 MHz) δ 1.18-1.35 (m, 6 H, 2 CH_3 , 2.04, 2.13, 2.18, 2.10-2.26 (3 s + m, 10 H, 3 Ac + H_{3a}), 2.67 (dd, J = $14.3 \text{ Hz}, 1 \text{ H}, \text{H}_{3e}$, 3.20 (br s, 1 H, 4-OH), 3.29 (ddd, $J = 9.5.2 \text{ Hz}, 1 \text{ H}, \text{H}_5$), 3.65 (brd, 1 H, 5-OH), 4.10-4.28 (m, 5 H, 2 CH₂ + H₄), 4.33 (dd, J = 12,5 Hz, 1 H, H₉), 4.46-4.58 (m, 2 H, H₆+H₉), 5.38 (dd, J = 5,1 Hz, 1 H, H₇), 5.46 (dt, J= 5,5,1.5 Hz, 1 H, H₈). ¹H NMR (d_6 -DMSO, 250 MHz) δ 1.08-1.25 (m, 6 H, 2 CH₃), 1.98, 2.01, 2.06, 2.03-2.15 (3 s + m, 10 H, 3 Ac + H_{3a}), 2.39 (dd, J =11,3,1 Hz, 1 H, H $_{3e}$), 3.17-3.26 (m, 1 H, H₅), 3.84-3.89 (m, 1 H, H₄), 3.97-4.18 (m, 5 H, 2 CH₂ + H₉), 4.40 (dd, J = 8,1.5 Hz, 1 H, H₆), 4.48 (dd, J = 9,2 Hz, 1 H, H₉), 4.76 (d, J = 6 Hz, 1 H, 5-OH), 4.93 (d, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H, 4-OH), 5.20 (ddd, J = 2 Hz, 1 H), 5.20 (ddd, J = 2 Hz), 5.20 (ddd, J = 2 Hz)J = 5.5, 3.5, 2 Hz, 1 H, H₈), 5.42 (dd, J = 3.5, 1.5 Hz, 1 H, H₇). ¹³C NMR (CDCl₃, 62 MHZ) & 13.41, 13.55 (2 CH₃ Et), 20.38, 20.65 (3 CH₃ Ac), 35.02 (C₃), 61.18, 61.92, 62.00 (2 CH₂ Et, C₉), 65.93, 66.11, 69.49, 69.99 (C_{4.5.6.7.8}), 77.77 (C₂) 166.83, 167.93, 169.96, 170.41, 171.63 (5 C=O).

Anal. Calcd for $C_{20}H_{30}O_{13}$: C, 50.21; H, 6.32; O, 43.47. Found: C, 50.05; H, 6.25; O, 43.48.

7,8,9-Tri-O acetyl-2-carbethoxy-2,3-dideoxy-4-oxo-D-Ethyl glycero-D-talo-non-2-ulosonate (35). Diol 34 (872 mg, 1.82 mmol) and dibutylstannyl oxide (498 mg, 2 mmol) were refluxed in toluene (50 mL) during 4 h with azeotropic removal of water. The solvent was then partially removed by distillation (until a ~ 2 mL volume was reached) and the mixture was cooled to room temperature. Anhydrous dichloromethane (2 mL) was added, and tributylstannyl methoxide (610 mg, 1.9 mmol) was added in one portion. A solution of bromine in dichloromethane (0.25 M, 8 mL, 2 mmol) was then added dropwise as long as decoloration occured. Concentration to dryness and flash-chromatography of the residue (1:1 AcOEt-hexane) allowed isolation of the pure hydroxyketone 35 (369 mg, 80%) as a pale yellow oil; [α]_D²⁰ -17° (c 2.8, CH₂Cl₂). ¹H NMR (CDCl₃, 250 MHz) δ 1.22-1.35 (m, 6 H, 2 CH₃), 2.08, 2.11, 2.20 (3s, 9 H, 3 Ac), 3.07 (dd, J = 14,1 Hz, 1 H, H_{3a}), 3.23 (d, J = 14 Hz, 1 H, H_{3e}), 3.58 (brd, J = 3.5 Hz, 1 H, 5-OH), 3.98 $(ddd, J = 10, 3.5, 1 Hz, 1 H, H_5), 4.09 (dd, J = 10, 1.5 Hz, 1 H, H_6), 4.18-4.36$ $(m, 5 H, 2 CH_2 + H_9), 4.50 (dd, J = 12, 2 Hz, 1 H, Hg), 5.40 (ddd, J = 6, 5, 2)$ Hz, 1 H, H₈), 5.57 (dd, J = 6,1.5 Hz, 1 H, H₇). ¹³C NMR (CDCl₃, 62 MHZ) δ 13.83 (2CH₃ CO₂Et), 20.65, 20.71 (3 CH₃ Ac), 43.46 (C₃), 61.87, 62.77, 62.99 (2CH₂ Et, C₉), 68.68, 69.61, 72.09, 77.19 (C_{5.6,7.8}), 82.32 (C₂), 165.49, 166.36, 169.90, 170.64, 171.63 (5 C=O), 203.29 (C₄).

Anal. Calcd for $C_{20}H_{28}O_{13}$: C, 50.42; H, 5.92; O, 43.66; Found: C, 50.89; H, 6.11; O : 43.60.

Ethyl 7,8,9-Tri-O-acetyl-2-carbethoxy-2,3-dideoxy-D-glycero-Dgalacto-non-2-ulosonate (36). The hydroxy ketone 35 (565 mg, 1.18 mmol) and cerium chloride (1.33 g, 3.56 mmol) were dissolved in methanol (5 mL) at room temperature. After a few minutes, sodium borohydride (76 mg, 2 mmol) was added and reduction occurred immediately. Acetic acid was then added dropwise until pH rose to neutrality. After dilution with water and extraction with ether (3 x 50 mL), the organic layer was dried over sodium sulfate and the solvent was evaporated, giving rise to a pale yellow oil which was shown to be a 4:1 mixture of *cis* and *trans* diols 34 and 36 (513 mg, 91%). Chromatography (AcOEt : hexane, 2:1) of the mixture gave, as the second fraction in order of elution, the pure diol 36; $[\alpha]_D^{20}$ -8° (*c* 1.8, CH₂Cl₂). 1H NMR (CDCl₃, 250 MHz) δ 1.23-1.34 (m, 6 H, 2 CH₃), 1.96 (dd, J = 13.5, 11.5 Hz, 1 H, H_{3a}), 2.05, 2.13, 2.21 (3s, 3 H each, 3 Ac), 2.73 (dd + brs, J = 13.5, 4 Hz, 2 H, H_{3e} + OH), 3.02 (brt, J = 9 Hz, 1 H, H₅), 3.57 (brs, 1 H, OH), 3.61 (ddd, J = 11.5, 8, 4 Hz, 1H, H₄), 3.83 (dd, J = 9.5, 1.5 Hz, 1 H, H₆), 4.14-4.30 (m, 4 H, 2 CH₂), 4.37 (dd, J = 12, 4 Hz, 1 H, H₉), 4.49 (dd, J = 12, 2 Hz, 1 H, H₉), 5.28 (dd, J = 7, 1.5 Hz, 1 H, H₇), 5.43 (ddd, J = 7.5, 4, 2 Hz, 1 H, H₈). ¹³C NMR (CDCl₃, 50 MHZ) δ 13.49, 13.63 (2 CH₃ Et), 20.38, 20.47, 20.62 (3 CH₃ Ac), 35.85 (C₃), 61.77, 61.87, 62.06 (2 CH₂ Et, C₉), 68.72, 69.42, 70.40, 74.25 (C_{4,5,6,7,8}), 80.92 (C₂), 166.28, 167.19, 169.82, 170.41, 171.77 (5 C=O).

Anal. Calcd for C₂₀H₃₀O₁₃: C, 50.21; H, 6.32; O, 43.4. Found: C, 50.36; H, 6.58; O, 43.58.

Methyl 4,5,7,8,9-Penta-O-acetyl-2,3-dideoxy-D-glycero-D-galactonon-2-ulosonate (37). Compound 36 (125 mg, 0.26 mmol) was treated with aq LiOH (1 M, 2 mL) for 1 d. After addition of 4 N aq HCl (until pH 1-2 was reached) the mixture was heated under reflux during 8 h, and then cooled to room temperature before readjusting the pH to 7-8 by adding 10% aq NaHCO₃. The solvent was removed under reduced pressure and DMF (2 mL) and $CH_{3}I$ (2 mL) were added. The mixture was stirred at room temperature for 2 d, concentrated to dryness, and then treated overnight at room temperature with a 1:1 (vol) mixture of Ac₂O-pyridine (8 mL). Coevaporation with toluene followed by flash-chromatography of the residue (1:2 AcOEthexane) gave 37 (90 mg, 73%). $[\alpha]_D^{30} = -51^{\circ}$ (c 2.9, CH₂Cl₂). ¹H NMR (CDCl₃, 250 MHz) & 2.02, 2.03, 2.05, 2.12, 2.17 (5 s, 15 H, 5 Ac), 1.95-2.10 (1 m, 1 H, H_{3ax}), 2.52 (ddd, J = 14, 4, 1.5 Hz, 1 H, H_{3eq}), 3.77 (s, 3 H, Me), 3.83 $(dd, J = 9.5, 1.5 Hz, 1 H, H_6), 4.17 (dd, J = 12, 4 Hz, 1 H, H_9), 4.29 (dd, J = 12)$ 10, 2 Hz, 1 H, H₆), 4.30 (dd, J = 12, 2 Hz, 1 H, H₉), 4.57 (br d, J = 6 Hz, 1 H, H_2), 4.82 (t, J = 10 Hz, 1 H, H_5), 4.96 (ddd, J = 10, 8, 4 Hz, 1 H, H_4), 5.32-5.43 (m, 2 H, H_{7.8}). ¹³C NMR (CDCl₃, 50 MHZ) δ 20.54, 20.65, 20.84, 21.01, 31.49, 52.32, 61.93, 66.64, 68.11, 68.24, 69.79, 71.29, 71.62, 169.71, 169.86,170.01, 170.14, 170.25, 170.65. IR (neat, cm⁻¹) 2955.9, 1749.7, 1436.7, 1371.0, 1225.5, 1151.9, 1118.1, 1057.7.

Anal. Calcd for C₂₀H₂₈O₁₃: C, 50.42; H, 5.92; O, 43.66. Found: C, 50.95; H, 5.96; O, 43.28.

Methyl 4,5,7,8,9-Penta-O -acetyl-2,3-dibromo-2,3-dideoxy-β-Derythro-L-manno-non-2-ulosonate (38). To a refluxing solution of 37 (120 mg, 0.25 mmol) in CCl₄ (10 mL) was added N-bromosuccinimide (2 g, excess). More NBS was added (1 g each day) in order to compensate bromine loss. After 5 d, the mixture was cooled to room temperature, filtered, and the filtrate concentrated. Flash-chromatography of the residue (1:2 AcOEthexane) gave **38** (120 mg, 76%); $[\alpha]_D^{30} \cdot 49^\circ$ (c 4.5, CH₂Cl₂). ¹H NMR (CDCl₃, 400 MHz) δ 2.02,-2.18 (5 s, 15 H, 5 Ac), 3.92 (s, 3 H, Me), 4.14 (dd, J = 12.5, 5 Hz, 1 H, H₉), 4.41 (dd, J = 12.5, 2 Hz, 1 H, Hg), 4.45 (dd, J = 9.5, 2 Hz, 1 H, H₆), 5.09 (d, J = 3 Hz, 1 H, H₃), 5.31 (ddd, J = 8, 5, 2 Hz, 1 H, H₈), 5.45 (dd, J = 8, 2 Hz, 1 H, H₇), 5.50 (t, J = 9.5 Hz, 1 H, H₅), 5.56 (dd, J = 9.5, 3 Hz, 1 H, H₄). ¹³C NMR (CDCl₃, 62 MHZ) δ 20.44, 20.52, 20.61, 20.68, 20.91, 52.46, 53.89, 61.65, 63.67, 65.82, 68.73, 69.65, 74.98, 91.08, 163.74, 169.14, 169.77, 170.56. IR (neat, cm⁻¹) 3058.8, 2955.9, 1753.9, 1436.8, 1371.1, 1266.0, 1230.1, 1068.4.

Anal. Calcd for C₂₀H₂₆Br₂O₁₃: C, 37.88; H, 4.13; Br, 25.20; O, 32.79. Found: C, 37.92; H, 4.21; Br, 24.57; O, 32.51.

Preparation of **39** and **42** directly from the mixture **5ab+6ab**. The same procedure as for preparation of **39** was applied to a mixture of adducts **5ab** and **6ab** (850 mg, 2.37 mmol), providing a mixture of 2,3-dideoxy derivatives (800 mg, 71%) that was treated by NBS, giving finally **39** + **42** (552 mg, 52%).

Anal. Calcd for C₂₀H₂₆Br₂O₁₃: C, 37.88; H, 4.13; Br, 25.20; O, 32.79. Found: C, 37.95; H, 4.35; Br, 25.31; O, 32.51.

Methyl 4,5,7,8,9-Penta-O -acetyl-2,3-dibromo-2,3-dideoxy-β-Derythro-L-altro-non-2-ulosonate (39). Starting from mixture of adducts 6ab (145 mg, 0.405 mmol), bishydroxylation following the same procedure as for preparation of compound 34 followed by Ac₂O-pyridine acetylation gave a mixture (157 mg, 82%). A portion of the mixture (145 mg, 0.3 mmol) of intermediate 2,3-dideoxy compounds, was submitted to the foregoing NBS oxidation procedure using 217 mg of NBS (1.2 mmol). Reflux during 1 d and usual work-up gave 39 (64 mg, 34%). ¹H NMR (CDCl₃, 250 MHz) δ 2.04, 2.06, 2.10, 2.11, 2.21 (5 s, 15 H, 5 Ac), 3.90 (s, 3 H, Me), 4.18 (dd, J = 12.5, 4Hz, 1 H, H₉), 4.45 (dd, J = 12.5, 2 Hz, 1 H, Hg), 4.57 (dd, J = 10, 2 Hz, 1 H, H₆), 4.92 (d, J = 2 Hz, 1 H, H₃), 5.33 (ddd, J = 7, 4, 2 Hz, 1 H, H₈), 5.50 (t, J =3 Hz, 1 H, H₄), 5.56 (dd, J = 7, 2 Hz, 1 H, H₇), 5.58 (dd, J = 10, 3 Hz, 1 H, H₅). ¹³C NMR (CDCl₃, 62 MHZ) δ 20.44, 20.66, 21.01, 46.92, 53.66, 61.12, 61.52, 66.22, 68.75, 69.53, 71.14, 88.66, 164.20, 169.25, 169.50, 169.69, 170.51. IR (neat, cm⁻¹) 3057.4, 2958.2, 1755.8, 1440.2, 1371.9, 1266.4, 1220.8, 1075.1.

Preparation of 42. Starting from mixture of adducts **5ab** (97 mg, 0.27 mmol), bishydroxylation following the same procedure as for preparation of compound **34** followed by Ac_2O -pyridine acetylation gave **40** (45 mg) and **41ab** (70 mg) in a global 90% yield. Equatorial isomer **41a** could be isolated from the mixture by further chromatography.

Methyl 4,5,7,8,9-Penta-O-acetyl-2,3-dideoxy-D-glycero-D-mannonon-2-ulosonate (40). ¹H NMR (CDCl₃, 200 MHz) δ 1.97, 1.98, 2.03, 2.05, 2.07 (5 s, 15 H, 5 Ac), 2.10-2.40 (m, 2 H, H_{3,3}'), 3.82 (s, 3 H, Me), 4.29 (dd, J = 9.5, 1 Hz, 1 H, H₆), 4.40 (dd, J = 12, 7 Hz, 1 H, H₉), 4.49 (dd, J = 12, 3 Hz, 1 H, H₉'), 4.51 (dd, J = 5, 1 Hz, 1 H, H₂), 4.93 (ddd, J = 11, 5, 3 Hz, 1 H, H₄), 5.21 (dd, J = 9.5, 2 Hz, 1 H, H₇), 5.23-5.27 (m, 1 H, H₅), 5.47 (ddd, J = 7, 3, 2 Hz, 1 H, H₈). ¹³C NMR (CDCl₃, 50 MHZ) δ 20.44, 20.55, 20.68, 20.73, 20.81, 29.58, 52.47, 61.62, 64.74, 66.49, 68.21, 70.34, 70.59, 72.25, 169.46, 169.92, 170.10, 170.41, 170.62, 170.91.

Methyl 4,5,7,8,9-Penta-O-acetyl-2,3-dideoxy-β-D-glycero-D-allonon-2-ulosonate (41a). [α]_D³⁰ 46° (c 0.6, CH₂Cl₂). ¹H NMR (CDCl₃, 250 MHz) δ 2.04, 2.05, 2.13, 2.15 (4 s, 15 H, 5 Ac), 1.96-2.23 (1 m, 2 H, H_{3,3}'), 3.87 (s, 3 H, Me), 4.03 (dd, J = 10, 2 Hz, 1 H, H₆), 4.22 (dd, J = 12, 7 Hz, 1 H, H₉), 4.37 (dd, J = 12, 2.5 Hz, 1 H, H₂), 4.64 (dd, J = 12, 2 Hz, 1 H, H₉), 4.92 (dd, J= 10, 3 Hz, 1 H, H₅), 5.23 (ddd, J = 7, 4, 2 Hz, 1 H, H₈), 5.38 (dd, J = 4, 2 Hz, 1 H, H₇), 5.56 (q, J = 3 Hzn 1 H, H₄). ¹³C NMR (CDCl₃, 50 MHZ) δ 20.38, 20.56, 20.66, 32.50, 53.35, 62.29, 65.95, 67.35, 70.07, 70.23, 71.25, 73.35, 169.45, 169.52, 169.78, 169.87, 170.44. IR (neat, cm⁻¹) 2957.8, 1745.9, 1439.3, 1373.0, 1223.1, 1146.4, 1061.1.

Anal. Calcd for C₂₀H₂₈O₁₃: C, 50.42; H, 5.92. Found: C, 50.22; H, 6.19.

Methyl 4,5,7,8,9-Penta-O -acetyl-2,3-dibromo-2,3-dideoxy-α-Derythro-D-altro-non-2-ulosonate (42). Mixture 41ab (320 mg, 067 mmol) was submitted to NBS oxidation (480 mg, 2.7 mmol) in refluxing CCl₄ (25 mL) during 6.5 h. Usual work-up gave 42 (310 mg, 73%). 42: $[\alpha]_D^{28}$ 55° (*c* 1.3, CH₂Cl₂). ¹H NMR (CDCl₃, 200 MHz) δ 2.03, 2.08, 2.11, 2.14, 2.20 (5 s, 15 H, 5 Ac), 3.91 (s, 3 H, Me), 4.27 (dd, J = 12, 5.75 Hz, 1 H, H9), 4.55 (dd, J =12, 2 Hz, 1 H, Hg), 4.63 (dd, J = 10.5, 2 Hz, 1 H, H6), 4.92 (d, J = 2.5 Hz, 1 H, H₃), 5.31 (dt, J = 2, 5.75 Hz, 1 H, H₈), 5.51 (dd, J = 5.75, 2 Hz, 1 H, H7), 5.57 (t, J = 2.5 Hz, 1 H, H4), 5.77 (dd, J = 10.5, 2.5 Hz, 1 H, H5). ¹³C NMR (CDCl₃, 62 MHZ) δ 20.40, 20.61, 20.79, 20.94, 47.06, 53.61, 61.86, 63.05, 68.59, 69.07, 69.62, 72.60, 87.70, 164.30, 169.06, 169.43, 169.74, 170.43. IR (neat, cm⁻¹) 2958.6, 1756.5, 1440.6, 1372.0, 1224.1, 1045.4.

Anal. Calcd for C₂₀H₂₆Br₂O₁₃: C, 37.88; H, 4.13; Br, 25.20; O, 32.79. Found: C, 38.82; H, 4.37; Br, 24.55; O, 32.17.

ACKNOWLEDGEMENTS

We thank C.N.R.S. and Université de Paris-Sud for financial support.

REFERENCES AND NOTES

- 1. Present address: Unité Mixte CNRS Béghin-Say "Sucrochimie" (UMR 143). c/o Eridania Béghin-Say CEI. B.P. 2132, 69603 Villeurbanne cedex, France.
- For reviews on use of aqueous media for organic reactions, see: A. Lubineau, J. Augé and Y. Queneau, Synthesis, 741 (1994); C. J. Li, Chem. Rev. 93, 2023 (1993); W. Blokzijl and J. B. F. N. Engberts, Angew. Chem. Int. Ed. Engl., 32, 1545 (1993).
- 3. A. Lubineau, J. Augé and N. Lubin, Tetrahedron Lett., 32, 7529 (1991).
- 4. A. Lubineau, J. Augé, E. Grand and N. Lubin, *Tetrahedron*, **50**, 10265 (1994).
- 5. A. Lubineau, J. Augé and N. Lubin, Tetrahedron, 49, 4639 (1993).
- 6. A. Lubineau and Y. Queneau, preceding paper in this issue.
- For examples of use of carbohydrates in cycloadditions reactions, see:
 R. M. Giuliano, Cycloadditions Reactions in Carbohydrate Chemistry; ACS Symposium Series no 494, American Chemical Society, Washington D. C., 1992, and references therein.
- 8. For syntheses towards 2-ulosonic acids, see bibliography in ref. 6.
- The same selectivity was observed starting from a glyceraldehyde based diene, either in regular or hetero Diels Alder reactions. See ref. 5 and A. Lubineau, J. Augé and N. Lubin, J. Chem. Soc., Perkin Trans. 1, 3011 (1990). For reviews on asymmetric Diels-Alder reactions, see : W. Oppolzer Angew. Chem. Int. Ed. Engl., 23, 876 (1984); L. A. Paquette, in Asymmetric Synthesis, Vol. 3 p 455, Academic Press, New York (1984); S. Masamune, W. Choy, J. S. Petersen and L. R. Sita, Angew. Chem. Int. Ed. Engl., 25, 1 (1985). For reviews on hetero Diels-Alder reactions, see: S. M. Weinreb and R. S. Staib, Tetrahedron, 38, 3087(1982) ; R. R. Schmidt, Acc. Chem. Res., 19, 250 (1986); J. Kametani and S. Hibino, Adv. Heterocycl. Chem., 42, 245 (1987).
- H. Paulsen, W. Bartsch and J. Thiem, Chem. Ber., 104, 2545 (1971); J. R. Hauske and H. Rapoport, J. Org. Chem., 44, 2472 (1979); N. Cohen, B. L. Banner, R. J. Lopresti, F. Wong, M. Rosenberger, Y. Y. Liu, E. Thorn and A. A. Liebman, J. Am. Chem. Soc., 105, 3661 (1983).

- A. Mukhopadhyay, S. M. Ali, M. Husain, S. N. Suryawanshi and D. S. Bhakuni Tetrahedron Lett., 30, 1853 (1989).
- Diene 1 appeared to be quite sensitive to acidic conditions; primary degradation product was (E, E)-hepta-2,4-diene-1,6,7-triol: ¹H NMR (CD₃OD, 200 MHZ) δ 3.40-3.55 (m, 2 H, H_{7,7'}), 4.03-4.28 (m, 3 H, H_{1,1',6}), 4.87 (br s, 3 H, 3 OH), 5.62-5.87 (m, 2 H, H_{2,5}), 6.10-6.38 (m, 2 H, H_{3,4}); ¹³C NMR (CD₃OD, 50 MHz) δ 63.09, 67.07, (C_{1,7}), 73.77 (C₆), 131.23, 131.99, 133.74 (C_{2,3,4,5}).
- B. Fraser-Reid, and B. Radatus, J. Am. Chem. Soc., 92, 5288 (1970); F. Gonzalez, S. Lesage, and A. S. Perlin, Carbohydr. Res., 42, 267 (1975);
 Y. K. Tam, and B. Fraser-Reid, Carbohydr. Res., 45, 29 (1975).
- 14. NMR data for **22** and **23** : (CDCl₃, 200 MHz): **22** δ 1.39, 1.40 (2 t, J = 7Hz, 3 H each, 2 CH₃), 2.47 (ddt, J = 18, 3.5, 2.5, 2.5 Hz, 1 H, H₃) 2.95 (ddd, J = 18, 5, 2.5 Hz, 1 H, H₃), 3.58 (t, J = 9 Hz, 1 H, H₇), 3.66 (t, J = 10.5 Hz, 1 H, H_{9ax}), 4.03 (ddd, J = 10.5, 9, 5.5 Hz, 1 H, H₈), 4.25, 4.26 (2q, J = 7 Hz, 2 H each, 2 CH₂), 4.38 (dd, J = 10.5, 5.5 Hz, 1 H, H_{9eq}), 4.50-4.62 (m, 1 H, H₆), 5.47 (s, 1 H, PhCH), 5.83-6.07 (m, 2 H, H_{4,5}), 7.30-7.55 (m, 5 H, Ph); **23** δ 1.22-1.38 (m, 6 H, 2CH₃), 2.55 (ddt, J = 18, 4, 3, 3 Hz, 1 H, H₃), 2.95 (ddd, J = 18, 5.5, 2.5, 1 H, H₃), 3.66 (dd, J = 10.5, 10 HZ, 1 H, H_{9ax}), 3.95 (dt, J = 9.5, 9.5, 5.5 Hz, 1 H, H₈), 4.08 (dd, J = 9, 4 Hz, 1 H, H₇), 4.19-4.40 (m, 5 H, 2 CH₂+H_{9eq}), 4.79-4.88 (m, 1 H, H₆), 5.51 (s, 1 H, PhCH), 5.87-6.14 (m, 2 H, H_{4,5}), 7.30-7.50 (m, 5 H, Ph).
- 15. The magnitude of the coupling constant between H-7 and H-8 provide consistent clues for differentiating the two types of conformations; see NMR data in relevant papers in ref. 8, and, among others: H. H. Brandstetter and E. Zbiral Liebigs Ann. Chem., 2055 (1983); E. Zbiral, and H. H. Brandstetter Monatsh. Chem., 116, 87 (1985); M. Salunkhe, M. Hartmann, W. Schmid and E. Zbiral, Liebigs Ann. Chem., 187 (1988); E. Zbiral, E. Schreiner, M. Salunkhe, G. Schulz, R. G. Kleineidam and R. Schauer, Liebigs Ann. Chem., 519 (1989).
- 16. S. David and A. Thieffry, J. Chem. Soc., Perkin Trans., 1 1568 (1979).
- 17. A. L. Gemal and J. L. Luche, J. Am. Chem. Soc., 103, 5454 (1981).
- Usefulness of 2,3-dibromo-2,3-dideoxy 2-ulosonic acid derivatives has been demonstrated. See for example: K. Okamoto, T. Kondo and T. Goto, Bull. Chem. Soc. Jpn., 60, 631 (1987). K. Okamoto, T. Kondo and T. Goto, Bull. Chem. Soc. Jpn., 60, 637 (1987); M. Nakamura, S. Fujita and H. Ogura, Chem. Pharm. Bull., 41, 21 (1993).
- 19. J. C. Sowden, J. Am. Chem. Soc. 72, 808 (1950).